



A Laser, a Genetic Algorithm and Some Molecules: Ingredients for Controlled Photchemistry

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Overview

- A review of laser power & effects on molecules
- Photochemical reactions with strong & weak field lasers
- Designing a laser pulse and using a genetic algorithm (GA) to control sequences of pulses
- Experimental results



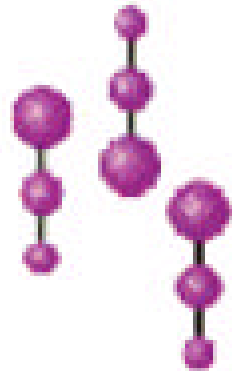
A Brief History of Laser Power

- The first lasers were continuous with powers of the order 10^3 W
- Q switching allows nanosecond pulses and powers up to 10^6 W
- Mode locking creates picosecond pulses and powers up to 10^9 W
- Further increases in power were achieved by chaining laser amplifiers
- Chirped Pulse Amplification achieves sub picosecond pulses and petawatt powers (10^{15} W)



The Effects of Power

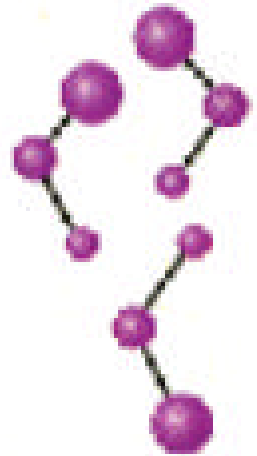
- Alignment



^aK. Yamanouchi, *Science*, **2002**, 295, 1659

The Effects of Power

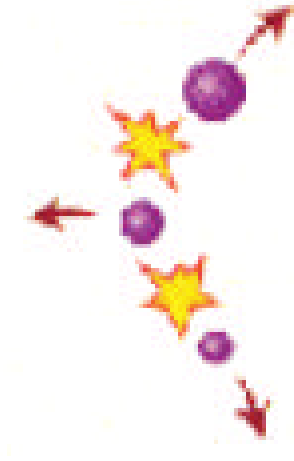
- Alignment
- Deformation



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The Effects of Power

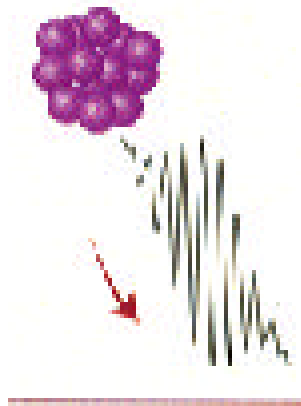
- Alignment
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The Effects of Power

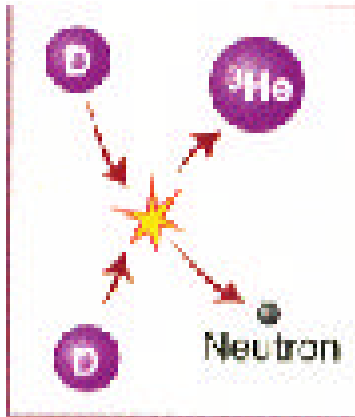
- Alignment
- Deformation
- Coulomb explosion
- X-ray emission



K. Yamanouchi, *Science*, **2002**, 295, 1659

The Effects of Power

- Alignment
- Deformation
- Coulomb explosion
- X-ray emission
- Nuclear reaction



K. Yamanouchi, *Science*, **2002**, 295, 1659



Dressing a PES

- Laser fields of 10^{14} to 10^{15} W/cm² can mix electronic states
- Effectively, the PES is deformed
- Controlling the deformation leads to controlling the direction of nuclear motion
- Selective bond breaking may be achieved



Photochemical Reactions

- Usually employs weak field lasers
- The laser must be tuned to the molecular resonance
- Limited to reactions where bond dissociation energies are less than ~ 50 kcal/mol
- Due to IVR, selective bond breaking is difficult
- We thus lose control over the reaction after initial irradiation



Strong Field Lasers

- Intensities are of the order of 10^{13} W/cm^2
- Leads to Stark shifting & multiphoton excitation
- Multiple eigenstates are brought into resonance
- Bandwidth restriction removed



Controlled Radiation Fields

- Such fields have been generated mainly in NMR experiments
- Various waveforms have been developed to:
 - Excite small regions of a spectrum
 - Suppress solvent peaks



Designing a Laser Pulse

- Designing a laser pulse for reaction control is more complex than modulating RF waves
- Requires knowledge of the *complete* Hamiltonian



Designing a Laser Pulse

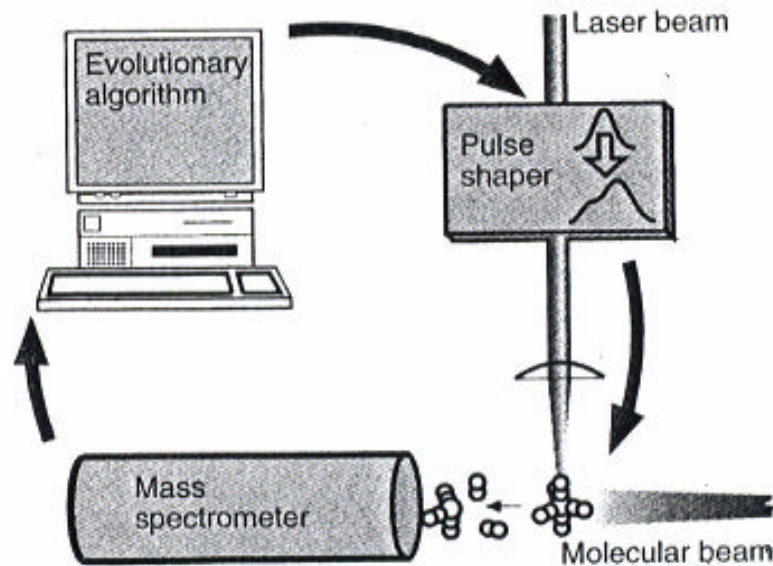
- Problems:
 - Intense laser fields cannot be treated as perturbations
 - Complete Hamiltonians are generally not available
 - The calculated field must account for errors in the Hamiltonian & actual beam
- Numerical design of a pulse is unfeasible



Closed Loop Methods

- An iterative method to reach an optimal state
 - Perturb the system
 - Observe the effect(s) of the perturbation
 - Modify the perturbation and repeat
- Stop when the system is in the required state
- This has been implemented using a GA as the controlling mechanism

Designing a Laser Pulse



- Solution - Analog Computation:
 - Use the laser itself as a *computer*!
 - Apply a field to the molecules
 - Analyze products
 - The report is then used to modify the laser field



Using a GA to Design Pulses

- Split a laser beam into 128 bands
- Each band can have its phase & amplitude modulated
- Modulation is achieved by mutations and crossover within a population of beams
- After modulation, the bands are recombined to give the unique pulse



The GA Details

- Groups of 16 neighboring pixels are tied together, giving a genome with 16 active sites
- Initial population consists of 40 randomly generated pulses
- Fields are propagated using proportional selection
- New fields are generated by cross over and mutation



Mechanisms of Control

- Two possible control mechanisms can occur:
 - Trivial Control
 - Non Trivial Control



Mechanisms of Control

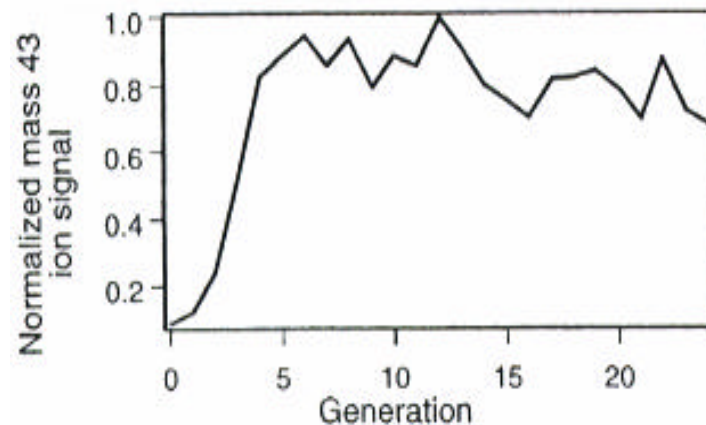
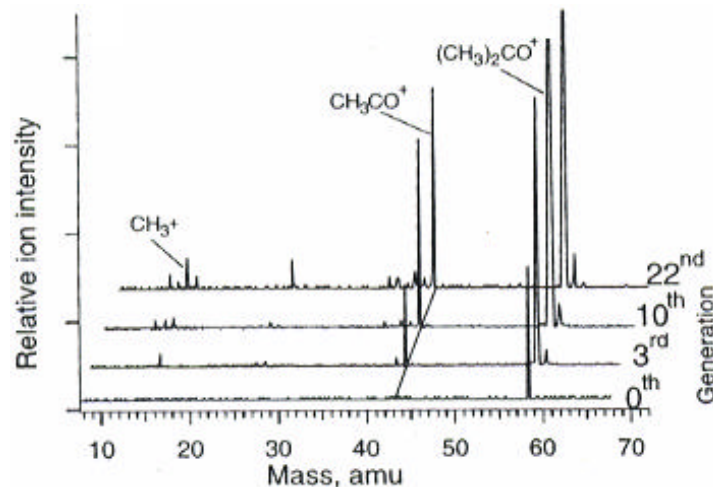
- Trivial Control
 - Due to simple intensity or pulse duration effects
- Non Trivial Control
 - Due to interaction of the laser pulse with the molecular wavepacket
 - Thus, dependant on the shape of the pulse



Mechanisms of Control

- The nature of control can be verified experimentally
- Ion yields from traditional and pulse shaping experiments are compared
- Pulse shaped experiments should not be significantly affected by pulse intensity or duration.

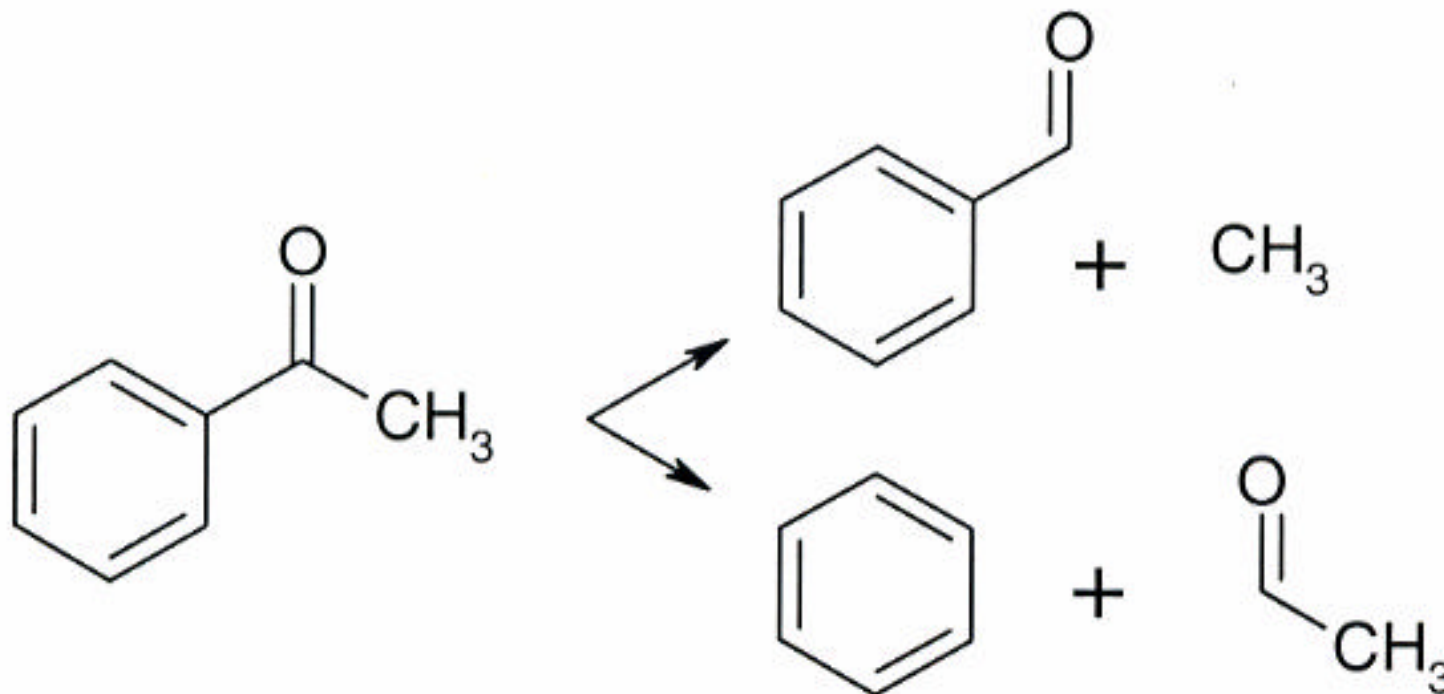
Catching the Signal



- The initial CH_3CO^+ signal is nearly invisible
- The GA is able to pick it out and hence maximize it
- The signal increases rapidly and then plateaus - variations are due to random change.

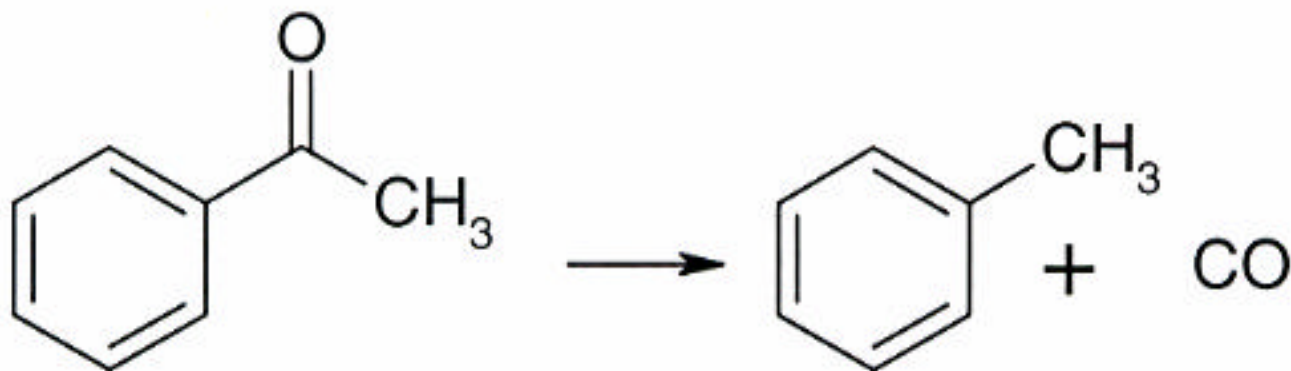
Acetophenone Reactions

- Acetophenone can undergo cleavage in two possible ways:



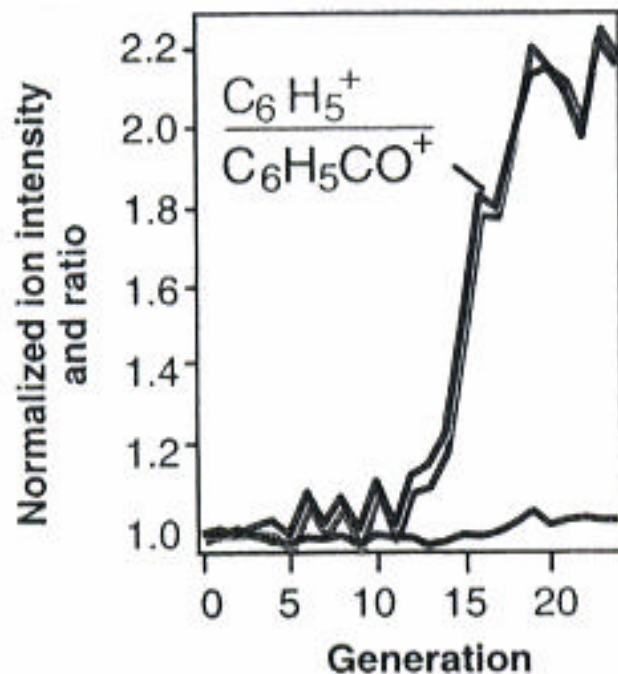
Acetophenone Reactions

- Acetophenone can also undergo a rearrangement to generate toluene



- It was possible to experimentally specify which path would be followed

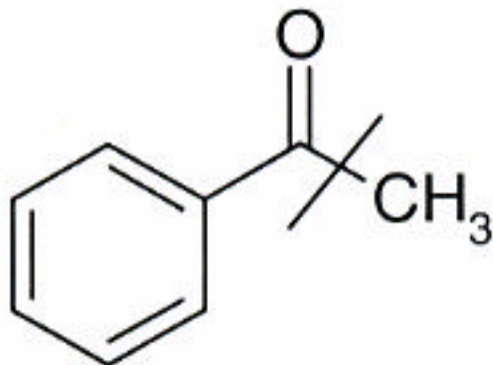
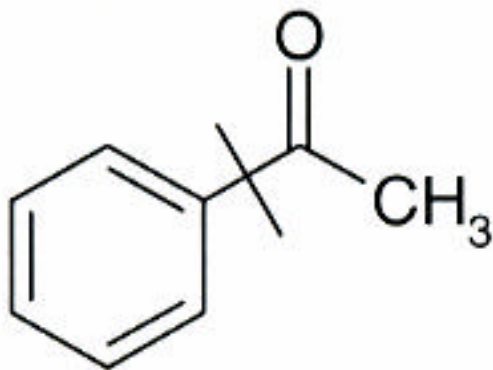
Dissociative Cleavages



- By setting the goal to maximization of the $C_6H_5^+ / C_6H_5CO^+$ ratio, we are specifying cleavage of the phenyl group
- The $C_6H_5CO^+$ signal remains relatively constant

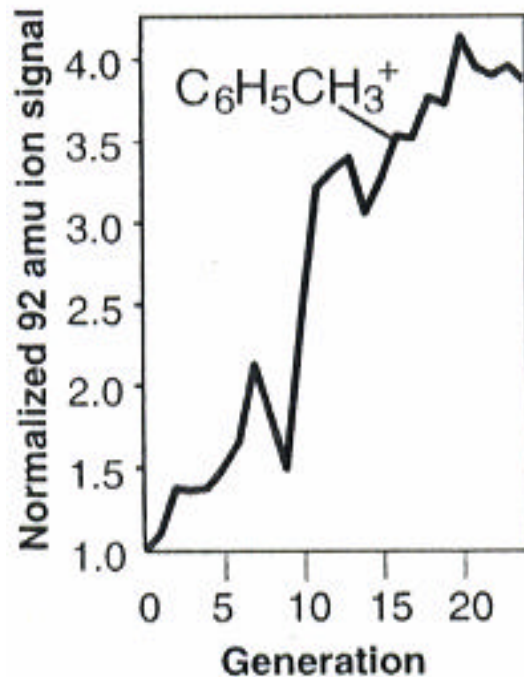
R.J. Levis et al, *Science*, **2001**, 292, 709

Interesting Features



- The Ph – COCH₃ bond energy is ~ 100 kcal/mol
- The CH₃ – COPh bond energy is ~ 85 kcal/mol
- We are thus able to guide a reaction along an energetically unfavorable pathway

Dissociative Rearrangement



- Tradition EI mass spectrum of acetophenone does not exhibit a toluene signal
- The TOF mass spectrum contains a toluene signal
- Hence, it is possible to instruct the GA to maximize the toluene pathway



Nature of Control

- In this case, control is non trivial:
 - Unlikely that trivial control could increase the phenyl intensity but keep that of phenylcarbonyl constant
 - Increasing pulse duration and lower intensities both decreased ion intensity monotonically in reference experiments.
 - No such correlation in pulse shaped experiments.



Conclusions

- A laser can indeed be *taught* to control the dynamics of a reaction
- *A priori* knowledge of the molecular PES or Hamiltonian is not required
- It is possible to drive a reaction in specific directions - cleavage or dissociation in this case
- Reactions with high dissociation energies can be optically controlled